Effects of experimental freezing on soil nitrogen dynamics in soils from a net nitrification gradient in a nitrogen-saturated hardwood forest ecosystem

Frank S. Gilliam, Adam Cook, and Salina Lyter

Abstract: This study examined effects of soil freezing on N dynamics in soil along an N processing gradient within a mixed hardwood dominated watershed at Fernow Experimental Forest, West Virginia. Sites were designated as LN (low rates of N processing), ML (moderately low), MH (moderately high), and HN (high). Soils underwent three 7-day freezing treatments (0, –20, or –80 °C) in the laboratory. Responses varied between temperature treatments and along the gradient. Initial effects differed among freezing treatments for net N mineralization, but not nitrification, in soils across the gradient, generally maintained at LN < ML ≤ MH < HN for all treatments. Net N mineralization potential was higher following freezing at –20 and –80 °C than control; all were higher than at 0 °C. Net nitrification potential exhibited similar patterns. LN was an exception, with net nitrification low regardless of treatment. Freezing response of N mineralization differed greatly from that of nitrification, suggesting that soil freezing may decouple two processes of the soil N cycle that are otherwise tightly linked at our site. Results also suggest that soil freezing at temperatures commonly experienced at this site can further increase net nitrification in soils already exhibiting high nitrification from N saturation.

Résumé : Nous avons étudié les effets du gel du sol sur la dynamique de N dans le sol le long d’un gradient de transformation de N dans un bassin versant dominé par des feuillus mélangés à la Forêt expérimentale de Fernow, en Virginie-Ocidentale. Les stations ont été classées sur la base du taux de transformation de N: LN (faible), ML (modérément faible), MH (modérément élevé) et HN (élevé). Les sols ont subi trois périodes de gel de 7 jours (0, –20 ou –80 °C) en laboratoire. Les réactions ont varié selon la température ainsi que le long du gradient. Les effets initiaux étaient différents selon l’intensité du gel dans le cas de la minéralisation nette mais pas dans le cas de la nitrification dans les sols le long du gradient qui s’est généralement maintenu à LN < ML ≤ MH < HN avec tous les traitements. Le potentiel de minéralisation nette de N était plus élevé à la suite d’un gel à –20 et –80 °C que chez le témoin et plus élevé qu’à 0 °C dans tous ces traitements. Le potentiel de nitrification nette suivait le même patron. La station LN faisait exception : la nitrification nette était faible peu importe le traitement. La réponse de la minéralisation de N au gel était très différente de celle de la nitrification, indiquant que le gel du sol peut découpler deux processus du cycle de N dans le sol qui sont par ailleurs étroitement reliés dans notre site. Les résultats indiquent aussi que le gel du sol à des températures qui surviennent couramment dans ce site peut augmenter davantage la nitrification nette dans les sols où la nitrification est déjà élevée à cause de la saturation en N.

[Traduit par la Rédaction]

Introduction

Among the paradoxes of global warming is a predicted increase in the likelihood of soils of north-temperate forests to freeze during winter months (Groffman et al. 2001a). This has important implications for the structure and function of forest ecosystems because freezing can alter several biogeochemical processes in affected forest soils, particularly those involved with N cycling in forest ecosystems. Freezing of soil can drastically influence the mobility and availability of N in forest soils by affecting microbial populations responsible for transforming unavailable organic N into mobile, available forms (Allen-Morley and Coleman 1989). This response is similar to that of drying–rewetting cycles (Skogland et al. 1988; DeLuca et al. 1992), which have been shown to substantially increase extractable NO₃ from nearly undetectable levels in N-limited soils (Gilliam and Richter 1985, 1988). A burst of microbial activity generally accompanies both thawing of frozen soil and rewetting of dried soil (Skogland et al. 1988).

In temperate forests, particularly those in more northern latitudes, snow cover acts seasonally as an effective insulator for soil by virtue of its high air content, mitigating the effects of low air temperature on soil, typically preventing...
soil from freezing and ensuring survival of a variety of microorganisms beneath snow cover (Schimel et al. 2004; Schmidt et al. 2009). Consequently, soils that are overlain by snowpack frequently go through winter months without freezing, despite extremes of low air temperatures (Lynch-Stieglitz 1994). Conversely, soils commonly freeze in forests of northern latitudes in the absence of snow. Working in hardwood and conifer forests of New Hampshire, Fahey and Lang (1975) found that the occurrence of soil freezing increased with elevation, varied with stand type, and lasted as late as May and June in that region.

Global warming has been causally connected to recent decreases in snowfall in north-temperate forests (Barnett et al. 2005). Long-term data from the Hubbard Brook Experimental Forest, New Hampshire, indicate that from 1955 to 2005, all measures of snowfall (i.e., depth, water content, and duration of cover) decreased significantly with time (Campbell et al. 2007). Thus, the frequency of soil freezing events is likely to increase in the future (Hardy et al. 2001).

Experimental approaches to investigate effects of freezing on soil have varied considerably in the literature. Morley et al. (1983) used extreme freeze–thaw cycles (~27 to 23 °C) in the laboratory that resulted in 40%–60% mortality of bacterial populations. Investigating freezing effects on soil food webs, Allen-Morley and Coleman (1989) subjected soil to –1 °C for 7 days and found that postfreezing recovery varied greatly among soil microbial populations. Cooke (1990) used liquid N2 to freeze contrasting soil types, finding that effects of freezing on nitrification enzyme activity varied greatly with type of soil.

In 1996, a snowpack manipulation experiment was established at Hubbard Brook Experimental Forest to examine the biogeochemical consequences of decreases in snowpack accumulation (Groffman et al. 2001a). This study found that even mild winters can bring about soil freezing in the absence (i.e., experimental removal) of snowpack (Hardy et al. 2001). It also determined that soil freezing increased mortality of fine roots (Tierney et al. 2001), increased soil NO3 without significantly altering net N mineralization and nitrification (Groffman et al. 2001b), and enhanced loss of N and P via soil solution (Fitzhugh et al. 2001).

Christopher et al. (2008) studied the effects of soil freezing on N mineralization and nitrification by sampling along a snow-cover gradient in Japan, sampling in contrasting soil types. They found simultaneous increases in N mineralization and decreases in nitrification in response to freezing, regardless of soil type. Austnes and Vestgarden (2008) used an experimental approach to examine the response of N in undisturbed soil columns from montane heathlands of southern Norway to freeze–thaw cycle and permanent frost treatments, finding that both treatments increased NH4 and decreased NO3. In another freeze–thaw experiment, Joseph and Henry (2008) found that N leaching increased nearly twofold following freezing and thawing of temperate old field soils.

This latter response, increased leaching of N in the form of NO3, has important relevance to effects of freezing in soils of N-saturated forest ecosystems wherein soil NO3 accumulates and is leached from soil into streams (Aber et al. 1998). This can be of serious concern to forest health if more frequent freezing events further increase NO3 leaching because it has been shown to deplete Ca and Mg availability in impacted soils (Peterjohn et al. 1996; Gilliam et al. 2001a, 2001b, 2005). We are aware of no studies that have examined effects of freezing on N dynamics in soils of N-saturated forests.

Several hardwood-dominated watersheds of the Fernow Experimental Forest (FEF), West Virginia, have been shown to be N saturated (Gilliam et al. 1996; Peterjohn et al. 1996), particularly the long-term reference watershed at FEF (WS4) (Stoddard 1994; Gress et al. 2007). Another symptom of N saturation, that of high relative nitrification (i.e., the percent of mineralized N converted to NO3), is especially evident at FEF. Long-term in situ (buried bag) incubations have revealed that nitrification is consistently close to 100% across three FEF watersheds, indicating that net N mineralization and nitrification are tightly linked at this site (Gilliam et al. 2001a, 2001b, 2004).

In addition, a notable degree of spatial heterogeneity in soil N dynamics has been described for WS4 as evidenced by spatial patterns of soil solution chemistry and in situ incubations (Peterjohn et al. 1999; Gilliam et al. 2001a, 2001b, 2005). Based on long-term data for soil water NO3 and in situ rates of net nitrification, we have identified a gradient of four sites within WS4 that vary in rates of NO3 production.

The purpose of this study was to examine the effects of soil freezing on net N mineralization and nitrification along a gradient of rates of N processing in an N-saturated central hardwood forest ecosystem. We subjected soil samples to three freezing temperature treatments, 0, –20, and –80 °C, addressing the following questions. (i) What are the initial effects of soil freezing on NH4 and NO3 production? (ii) What are the effects of soil freezing on net N mineralization and nitrification potentials and relative nitrification? (iii) How do these responses vary along the nitrification gradient?

Materials and methods

Study site

The study site, FEF, occupies ~1900 ha of the Allegheny Mountain section of the unglaciated Allegheny Plateau in Tucker County, West Virginia, adjacent to the Monongahela National Forest (39°03’N, 79°49’W). Averaging approximately 1430 mm-year1, precipitation at FEF varies seasonally and with elevation, generally greater during the growing season and at higher elevations (Gilliam et al. 1996). Long-term records at FEF indicate that, for the decade of 1991–2000, there were >1100 days with ambient temperatures <0 °C. Of these, nearly 20% were less than −10 °C. Snowpack typically ranges from a few days to <2 weeks (Adams et al. 1994). Thus, the likelihood of freezing is notable for this site.

Four sample sites were located in WS4, the long-term reference watershed at FEF that supports >100-year-old mixed-aged hardwood stands (Fig. 1). Soils at all sample sites are predominantly coarse-textured Inceptisols (loamy-skeletal, mixed mesic Typic Dystrochrept) of the Berks and Calvin series, sandy loams derived from sandstone. Soil pH generally varies between 3.50 and 4.00 (Gilliam et al. 2004). The woody overstory is characterized by a high diversity of tree species, with dominant species including sugar maple...
Field sampling

Sampling was done at four sites that were shown by previous investigations to represent a gradient in rates of net nitrification. Determination of this gradient was based on long-term soil water NO₃ data and 6 years of monthly (growing season) rates of net N mineralization and nitrification measured in situ from 1993 to 2007 and reported in part (growing season) rates of net N mineralization and nitrification. Determination of this gradient was based on previous investigations to represent a gradient in rates of net nitrification (LN), medium-low nitrification (ML), medium-high nitrification (MH), and high nitrification (HN).

Fig. 1. Map of WS4 of Fernow Experimental Forest, West Virginia, showing sample sites for the present study. Sites are as follows: low nitrification (LN), medium-low nitrification (ML), medium-high nitrification (MH), and high nitrification (HN).

(Acer saccharum Marsh.), black cherry (Prunus serotina Ehrh.), and northern red oak (Quercus rubra L.). The herbaceous layer of these sites comprises species typical of montane eastern deciduous forests, including violets (Viola spp.), blackberry (Rubus spp.), stinging nettle (Laportea canadensis (L.) Wedd.), and several species of ferns (Gilliam 2007).

Experimental treatments

Subsamples of soil from each plot were extracted for analysis of NH₄⁺ and NO₃⁻ immediately upon return to the laboratory (see Laboratory analyses below). Approximately 50 g of each sample was placed into each of three 120 mL sterile polyethylene Whirl-Pac bags for freezing treatment as follows: 0, –20, and –80 °C. The 0 and –20 °C treatments were administered by placing bags in separate Fisher Iso-temp incubators, which held soil temperature constant at the respective settings. The –80 °C treatment was administered by placing bags in a Thermo Electron Corporation freezer and was chosen as an extreme treatment to determine whether the intermediate treatment (–20 °C) represents a threshold temperature for freezing effects (i.e., slight or no differences between –20 and –80 °C treatments would suggest a –20 °C threshold). The remaining soil was kept in the original bag and refrigerated at 4 °C as control. All treated samples were subjected to treatments for 7 days.

Initial effects of freezing on soil N were assessed in this study to further separate the direct biotic and abiotic effects of freezing. These effects were assessed by extracting all treated samples immediately following thawing for 24 h (note: the 0 °C soils did not freeze to hardness due to the freezing point depression effect of soil solutes but were allowed to come to room temperature for the same time period as for the –20 and –80 °C samples). Effects of freezing treatment on net N mineralization and nitrification potential were assessed by incubating the remaining soil in all bags for 7 days at 25 °C prior to additional extraction.

Laboratory analyses

Soil moisture was monitored for all subsamples before and during incubation, and neither varied among samples or through time of incubation. Extraction and analysis for NH₄⁺ and NO₃⁻ followed methods described in Gilliam et al. (2005). Briefly, moist soils were extracted with 1 mol/L KCl at an extract to soil ratio of 10:1 (v/m); these were expressed as dry mass using a dry to moist conversion ratio from dried subsamples. Extracts were analyzed colorimetrically for NH₄⁺ and NO₃⁻ with an AutoAnalyzer 3 automatic analysis system. Net N mineralization rates (in μg N·g⁻¹ soil⁻¹·day⁻¹) were calculated as postincubation NH₄⁺ and NO₃⁻ minus preincubation NH₄⁺ and NO₃⁻. Nitrification rates (in μg N·g⁻¹ soil⁻¹·day⁻¹) were calculated as postincubation NO₃⁻ minus preincubation NO₃⁻. Relative nitrification was expressed as a percentage and calculated as net nitrification divided by net N mineralization multiplied by 100.

Data analyses

The effects of freezing on net N mineralization and nitrification were assessed as a function of freezing treatment by a two-way (treatment × temperature) ANOVA. Post hoc analyses were performed using a paired t-test (i.e., 0 vs. –20, and –20 vs. –80 °C). Statistical significance was set at α = 0.05.
fication were assessed by comparing means of control soils with those of soils treated at 0, –20, and –80 °C with one-way analysis of variance and least significant difference tests (Zar 1999). We considered initial effects of freezing to be any treatment effect found for extractions made immediately after freezing treatment. Effects of freezing on potential net N mineralization and net nitrification were said to be those found for extractions made following the 1-week incubation of all soils at 25 °C. Potential effects of soil weathering on net nitrification were assessed via correlation between mean nitrification and mean soil clay content across all four sites.

Results and discussion

Initial effects of freezing on net N mineralization and nitrification

Laboratory freezing exerted pronounced initial effects on net N mineralization, although this effect varied significantly \( (P < 0.05) \) among sites (Fig. 2). Initial effects differed significantly among freezing temperatures for net N mineralization and net nitrification were said to be those found for extractions made following the 1-week incubation of all soils at 25 °C. Potential effects of soil weathering on net nitrification were assessed via correlation between mean nitrification and mean soil clay content across all four sites.

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Initial effects of freezing on net N mineralization and nitrification

Laboratory freezing exerted pronounced initial effects on net N mineralization, although this effect varied significantly \( (P < 0.05) \) among sites (Fig. 2). Initial effects differed significantly among freezing temperatures for net N mineralization and net nitrification, but not nitrification, in soils across the gradient, which was generally maintained for all freezing treatments (i.e., rates of N processing at LN < ML ≤ MH < HN). An exception to these patterns was evident for the LN site wherein net nitrification remained very low across all freezing treatments (Fig. 2). Net N mineralization was highest at –80 °C for the higher N sites and was significantly lowest across all sites at 0 °C.

Initial responses of soil N to freezing (i.e., immediately after freezing treatment) were assessed in this study because they should exemplify the direct biotic and abiotic effects of freezing. Biotic effects include responses of microbial populations associated in soil N dynamics, which may initially decrease due to lysing of cells from ice formation in the cytosol, simultaneously increasing \( \text{NH}_4 \) and \( \text{NO}_3 \) that is released from the cytosol (Allen-Morley and Coleman 1989). Also, \( \text{NH}_4 \) production following cell lysis is possible through extracellular deaminase activity (Abdel-Fatah et al. 2003). Abiotically, freeze–thaw cycles facilitate physical

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**Table 1.** Characteristics of sample sites along a net nitrification gradient within WS4 of Fernow Experimental Forest, West Virginia.

<table>
<thead>
<tr>
<th>Characteristic</th>
<th>LN</th>
<th>ML</th>
<th>MH</th>
<th>HN</th>
</tr>
</thead>
<tbody>
<tr>
<td>Elevation (m)</td>
<td>808</td>
<td>838</td>
<td>838</td>
<td>833</td>
</tr>
<tr>
<td>Clay (%)</td>
<td>12.0</td>
<td>9.6</td>
<td>9.6</td>
<td>7.2</td>
</tr>
<tr>
<td>Sand (%)</td>
<td>69.8</td>
<td>66.3</td>
<td>70.3</td>
<td>66.8</td>
</tr>
<tr>
<td>Silt (%)</td>
<td>18.2</td>
<td>24.1</td>
<td>20.1</td>
<td>26.0</td>
</tr>
<tr>
<td>Organic matter (%)</td>
<td>17.0</td>
<td>17.3</td>
<td>13.8</td>
<td>17.0</td>
</tr>
<tr>
<td>In situ N mineralization (g N m(^{-2}) month(^{-1}))</td>
<td>0.1</td>
<td>2.2</td>
<td>2.9</td>
<td>4.3</td>
</tr>
<tr>
<td>In situ nitrification (g N m(^{-2}) month(^{-1}))</td>
<td>0.02</td>
<td>1.6</td>
<td>2.3</td>
<td>4.2</td>
</tr>
<tr>
<td>Soil water ( \text{NO}_3 ) (mg N L(^{-1}))</td>
<td>0.1</td>
<td>0.2</td>
<td>1.1</td>
<td>3.1</td>
</tr>
<tr>
<td>Foliar N (%)</td>
<td>1.64</td>
<td>2.38</td>
<td>2.34</td>
<td>2.55</td>
</tr>
<tr>
<td>Litter N (%)</td>
<td>1.30</td>
<td>1.63</td>
<td>1.65</td>
<td>1.81</td>
</tr>
<tr>
<td>Total basal area (m(^2) ha(^{-1}))</td>
<td>31.8</td>
<td>22.3</td>
<td>36.8</td>
<td>35.3</td>
</tr>
<tr>
<td>Total density (stems ha(^{-1}))</td>
<td>1025</td>
<td>825</td>
<td>1075</td>
<td>500</td>
</tr>
<tr>
<td>Dominant tree species</td>
<td>Quercus rubra</td>
<td>Quercus alba</td>
<td>Quercus rubra</td>
<td>Prunus serotina</td>
</tr>
<tr>
<td>Herb layer cover (%)</td>
<td>17.4</td>
<td>18.0</td>
<td>11.2</td>
<td>27.4</td>
</tr>
<tr>
<td>Dominant herb species</td>
<td>Smilax rotundifolia</td>
<td>Laportea canadensis</td>
<td>Smilax rotundifolia</td>
<td>Laportea canadensis</td>
</tr>
</tbody>
</table>

**Note:** Sites: low nitrification (LN), medium-low nitrification (ML), medium-high nitrification (MH), and high nitrification (HN). N data taken from ongoing work initially reported in Gilliam et al. (2001a, 2001b).

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**Fig. 2.** Initial effects of freezing on (a) net N mineralization and (b) net nitrification of soils from four sites within WS4 of the Fernow Experimental Forest, West Virginia. Experimental treatments are as follows: low freeze (T0, freezing at 0 °C for 7 days followed by incubation), medium freeze (T-20, freezing at –20 °C followed by incubation), and high freeze (T-80, freezing at –80 °C followed by incubation). Shown are means ± 1 SE of the mean. Means with the same letter (x, y, z) are not significantly different \( (P < 0.05) \) between sites for a given treatment. Means with same letter (a, b, c) are not significantly different \( (P < 0.05) \) between treatments for a given site.

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breakdown of soil organic matter, increasing surface area to volume ratios of soil organic materials and increasing accessibility to microbes.

**Effects of freezing on net N mineralization and net nitrification potentials**

Net N mineralization potential (hereafter, net N mineralization) (i.e., following a 7-day incubation at 25 °C) was significantly higher following freezing at –20 and –80 °C than the control, which was also significantly higher than the 0 °C treatment. A similar pattern was observed for net nitrification potential (hereafter, net nitrification), i.e., rates following –20 °C ≈ –80 °C > control > 0 °C (Fig. 3). Also, means among sites generally maintained the gradient pattern based on field data. That is, net nitrification was consistently highest at HN, lowest at LN, and intermediate at the ML and MH sites. More specifically, net nitrification at the LN site was consistently one to two orders of magnitude lower, ranging from 0.01 to 0.05 μg N g soil⁻¹ day⁻¹, than at the HN site, which ranged between ~3.0 and 5.0 μg N g soil⁻¹ day⁻¹ (Fig. 3).

Thus, whereas freezing treatments that resulted in a truly frozen state (i.e., –20 and –80 °C) increased net N mineralization and nitrification relative to controls, these processes were significantly inhibited at 0 °C. This may have arisen from two mechanisms. First, the 0 °C treatment may have suppressed N mineralization but not N immobilization, thus increasing the level of net N immobilization. Second, populations of ammonifying microbes and nitrifying bacteria may have decreased at 0 °C in a way that limited their response to the increasing temperature of incubation (25 °C). Significant increases in both net N mineralization and nitrification at –20 and –80 °C suggest that microbial cellular constituents (from cell lysis) and increased surface area of soil organic matter from the physical action of ice provided sufficient substrate to allow these populations to overcome low-temperature limitations.

It is of great interest that there were no significant differences in net N mineralization and nitrification between the –20 and –80 °C treatments. Generally outside the realm of surface temperatures of the biosphere (lowest natural temperature was –89.2 °C recorded 21 July 1983 at the Russian Vostok Station in Antarctica), the –80 °C treatment was used to test whether further decreasing temperature in a fully frozen state would have a proportional effect on soil N dynamics. Our results suggest that –20 °C may represent a threshold beyond which further decreases in temperature bring about minimal changes in microbial N processing. Clearly, freezing at –80 °C does not bring about notable declines in nitrifier populations, such as Nitrosomonas and Nitrobacter, at least not in soils with highly active populations, such as those that are predominant at the HN site (Gilliam et al. 2001a, 2001b).

Results vary considerably among studies in the literature that examine effects of freezing on soil N. Harding and Ross (1964) found that thawing after freezing at –20 °C increased net N mineralization and nitrification in some, but not all, soils examined, concluding that moisture and C contents of soils may influence this response. The field-based snow manipulation study at Hubbard Brook Experimental Forest found that freezing (brought about by experimentally removing snowpack) resulted in increases in soil NO₃ and enhanced loss of N in soil solution, but did so without significantly influencing net nitrification (Fitzhugh et al. 2001; Groffman et al. 2001a, 2001b). Joseph and Henry (2008) found that experimental freeze–thaw cycles of temperate old field soils increased leaching of N nearly twofold. By contrast, Hentschel et al. (2009) concluded that moderate soil frost did not stimulate solute losses of N, dissolved organic C, and mineral ions from soils of montane Norway spruce (Picea abies (L.) Karst.) stands around Fichtelgebirge, Germany.

The degree to which net N mineralization and net nitrification are coupled (sensu Christopher et al. 2008) can be assessed by calculating relative nitrification. Robertson (1982) used relative nitrification as an independent test of site factors (e.g., soil pH, C to N ratio) that potentially influence in situ nitrification. Relative nitrification has been shown to vary considerably with stand type, with much higher relative nitrification generally associated with hardwood than with conifer forests (Bonilla and Rodé 1992; Gilliam et al. 2004; Fenn et al. 2005). High relative nitrification also has been...
linked to high rates of N deposition (Fenn et al. 2005) and N saturation (Gilliam et al. 2004).

Mean relative nitrification for the control treatment confirms earlier findings that net nitrification can be as high as 100% of net N mineralization in these soils (Gilliam et al. 2001a, 2001b, 2004). Mean values varied significantly among sites, however, in the following order: HN (97%) > ML (77%) > MH (61%) > LN (7%), with the general effect of freezing being to decrease relative nitrification (Fig. 4). All freezing treatments significantly decreased relative nitrification at the LN site. Freezing at –20 and –80 °C, but not at 0 °C, caused significant decreases at the ML and MH sites, whereas –20 °C (but neither 0 nor –80 °C) significantly decreased relative nitrification at the HN site (Fig. 4). Therefore, we conclude that soil freezing may serve to decouple two processes of the soil N cycle that can otherwise be tightly linked at our site, similar to the conclusions of Christopher et al. (2008).

Gradient factors
It is notable that the gradient pattern in net nitrification based on long-term in situ incubations (Table 1) was generally maintained under the controlled conditions in the laboratory (Figs. 2b and 3b). This strongly suggests that the gradient patterns observed from the field have arisen from variation in soil microbial communities (i.e., composition and activity) of the respective sites rather than the more transient ambient factors, such as moisture and temperature, that are otherwise important in controlling microbial processes (Gilliam et al. 2001a, 2001b). Accordingly, we were interested in determining which, if any, site variables, many of which are listed as “Characteristics” in Table 1, might best explain this gradient pattern. Of these, soil clay content was correlated most closely with net nitrification across the four sites, independent of treatment (Fig. 5). Furthermore, there is considerable evidence to suggest that this gradient is the product of differential rates of mineral weathering during the process of soil formation (Jenny 1980). For example, although soils of WS4 are underlain by the same parent material (Gilliam et al. 2005), Tajchman et al. (1988) demonstrated that over a 35-year period (1948–1982), net radiation, a principle driver in weathering of primary and secondary soil materials (Jenny 1980; Rech et al. 2001), was highest on southwest-facing upper slopes of the watershed and lowest on the north- and northeast-facing slopes (see Fig. 1).

Although it is highly speculative, results from our study suggest that, at least across the 35 ha scale of WS4, weathering may represent an ultimate control on nitrification. More highly weathered soils typically have higher clay content (Jenny 1980), and clay content was negatively correlated with net nitrification among our samples site, a relationship that was not substantially altered by freezing treatment (Fig. 5). This is consistent with conclusions of Gilliam et al. (2005), a study that included the LN and HN sites as part of a different gradient study at FEF. They found that exchangeable Al, another indicator of weathering status of soil (Troeh and Thompson 2005), was highest at LN (713 μg Al g⁻¹ soil) and lowest at HN sites (149 μg Al-g soil⁻¹). Although our work was carried out on a much smaller spatial scale, these conclusions are also consistent with those of Reich et al. (1997), who examined factors affecting net N mineralization across 50 forest stands at six sites in Wisconsin and Minnesota. They discovered that soil texture was more important than stand type in explaining spatial patterns of N mineralization.

Implications for soil freezing in an N-saturated hardwood forest ecosystem
The variety of responses of soil N dynamics to freezing, as shown in both field- and laboratory-based studies, precludes broad generalizations or predictions on effects of soil freezing in forest ecosystems. Although some of this arises from variation among studies in methodology used, much of it is likely the result of (i) intersite variation in N status
and soil microbial communities, (ii) site-specific factors that most profoundly influence those communities, and (iii) variable microbial response to freezing. Results of our study, a laboratory-based experiment that employed a field-based gradient in net nitrification that was largely maintained regardless of experimental treatment, suggest strongly that this is the case, at least at the 35 ha scale of a hardwood-dominated watershed. The intermediate temperatures used in this study (0 and –20 °C) are not uncommon for ambient conditions at FEF. Long-term data from the Timber and Watershed Laboratory (USDA Forest Service) indicate that there were 1134 days <0 °C over the decade of 1991–2000, during which time snowpack typically lasts a few days at a time (Adams et al. 1994).

One of the symptoms of N saturation is an increase in the predominance of net nitrification (Aber et al. 1998), and WS4 has been cited as one of the better examples of an N-saturated ecosystem (Stoddard 1994; Peterjohn et al. 1996; Gress et al. 2007). Our results suggest that soil freezing at temperatures commonly experienced at this site can further increase net nitrification in soils already exhibiting high nitrification from N saturation, potentially exacerbating problems associated with N saturation, such as decreased growth rates of dominant hardwood species, which has been demonstrated at FEF (May et al. 2005; DeWalle et al. 2006).

Another characteristic of N saturation is high relative nitrification. That is, as N status increases, the amount of N mineralized from organic to inorganic forms that eventually become nitrified increases (Gilliam et al. 2001a, 2001b, 2004; Fenn et al. 2005), a pattern also supported by our data. However, net N mineralization increased in response to freezing to a degree greater than net nitrification at all sites other than HN. Thus, response of N mineralization to freezing may differ greatly from that of nitrification, suggesting that soil freezing may serve to decouple two processes of the soil N cycle that are otherwise tightly linked at our site.

Finally, because our sample sites represent discrete areas (i.e., subcatchments) within WS4, the N gradient represents a degree of spatial heterogeneity of N availability within the watershed. Thus, it is possible, based on results shown in Fig. 3a wherein there were no longer significant differences in net N mineralization among sites at –20 °C, that one of the effects of freezing at temperatures quite common to the region may be to decrease the spatial heterogeneity of N mineralization in the watershed. This is a response with relevance to forest biodiversity. For example, Gilliam (2007) found that up to 90% of plant biodiversity of temperate forest ecosystems is found in the herbaceous layer, a vegetation stratum that is sensitive to spatial and temporal variation in soil N availability. It has been suggested that decreases in spatial heterogeneity of soil N can lead to decreases in herb layer diversity (the “N homogeneity hypothesis”, see Gilliam 2006; Bobbink et al. 2010). Accordingly, the effects of soil freezing might have direct implications for biodiversity in forest ecosystems.

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References


