

Lecture 10

The Nitrogen (N) Cycle

- Atmosphere is ca. 78 % N₂ but most is unavailable to living things because ...
- N is important because ...
- Microbial processes are important in the steps of the N cycle.

The nitrogen cycle has 5 basic steps


1) Nitrogen Fixation: N₂ => NH₃

Root Nodules on a Legume



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Heterocysts in *Anabaena*




A black and white micrograph showing a chain of spherical cells of the cyanobacterium *Anabaena*. One cell is significantly larger and more rounded than the others, labeled 'H' for heterocyst. Two smaller cells are labeled 'PB' for protein bodies. A red arrow points from the word 'Heterocyst' to the 'H' cell. The letter 'b' is in the bottom left corner.

Heterocyst

Humans Fix Nitrogen Too !


The Haber Process



Fritz Haber

$$4\text{N}_2 + 12 \text{H}_2 + \text{catalyst} \rightleftharpoons 8 \text{NH}_3$$

at 500°C & several hundred atmospheres of pressure



A photograph of an industrial facility, likely a Haber process plant, with several large towers and pipes, situated in a dry, open landscape.

2) Ammonification: organic N \Rightarrow NH_3

3) Nitrification: $\text{NH}_3 \Rightarrow \text{NO}_2^- \Rightarrow \text{NO}_3^-$

2-step process - each step by different bacteria.

Step 1: oxidation of ammonia (NH_3) to nitrite (NO_2^-) by _____

Step 2: oxidation of nitrite (NO_2^-) to nitrate (NO_3^-) by _____

Both steps couple E-releasing oxidations to fixation of carbon - chemoautotrophs.

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If plants are called photoautotrophs, what would you call *Nitrosomonas*?

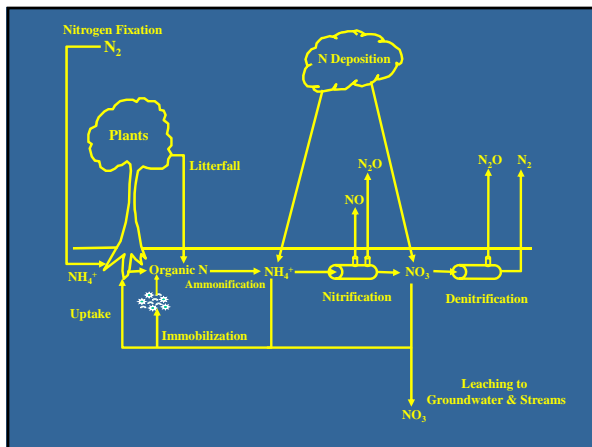
Would it help if you knew that

photo means they get E from light

auto means they get C from inorganic sources

4) Nitrogen Assimilation: $\text{NH}_3 \Rightarrow$ organic N
 $\text{NO}_3^- \Rightarrow$ organic N

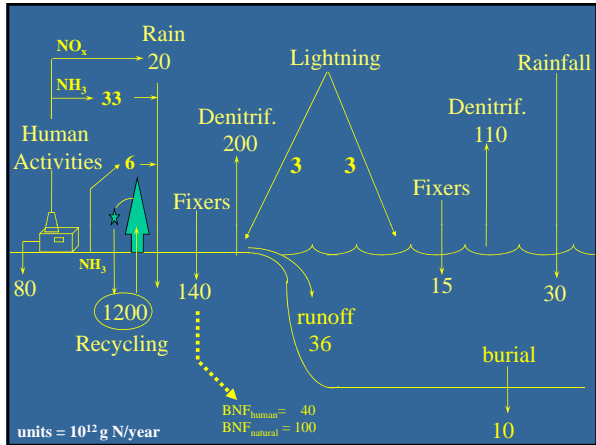
5) Denitrification: NO_3^- or $\text{NO}_2^- \Rightarrow \text{N}_2$ or N_2O

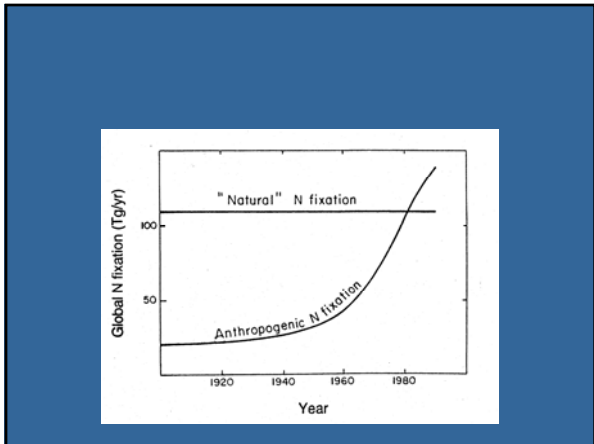


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Active N Pools

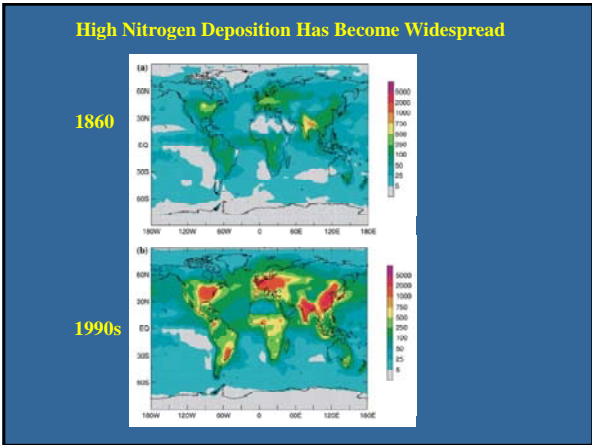
- Atmosphere $3,800,000 \times 10^{15}$ g N
- Ocean $21,000 \times 10^{15}$ g N
- Soil Organic Matter 95×10^{15} g N
- Terrestrial Biota 3.5×10^{15} g N





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Terrestrial Ecosystems Can be Overfertilized



- ### Potential Consequences of N Saturation
- Increased surface-water NO_3^- concentrations.
 - Enhanced losses of nutrient cations.
 - Soil acidification & greater soluble Al .

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Nitrogen Cycle Mean Residence Times

- Atmosphere
ca. 9 million years
- Land biota
ca. 3 years

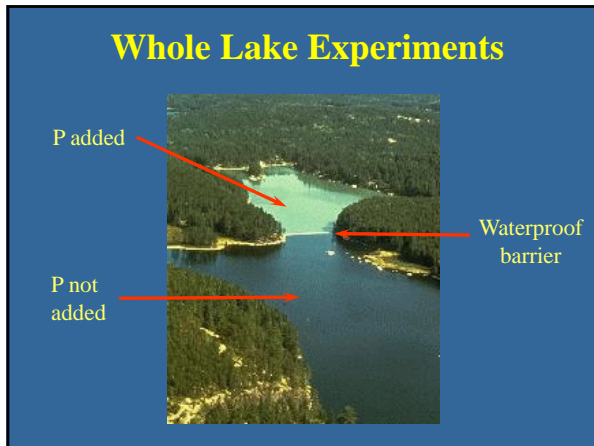
Summary of N Cycle

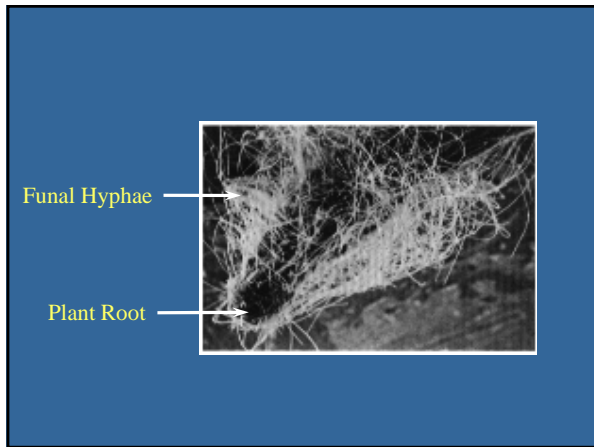
- Largest active pool = N_2 in atmosphere which is 181x > amount in ocean
- N in soil organic matter is 27x > amount in terrestrial biota
- Largest flux = uptake by plants of which almost all is from recycled organic N
- Human activities \approx 60 % of total inputs to land
- River flow \approx 40% total inputs to oceans

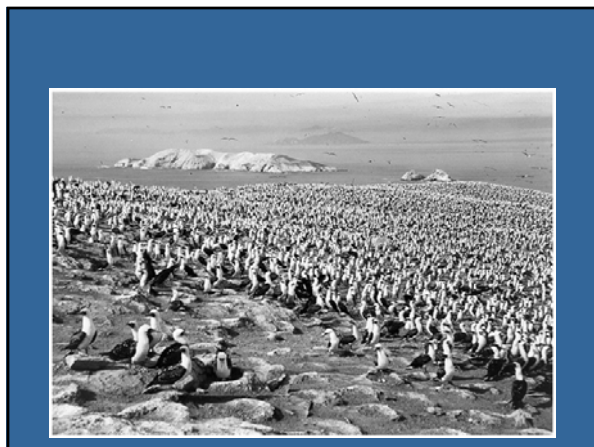
The Phosphorus (P) Cycle

- Example of a sedimentary cycle => no gaseous phase
- P is abundant in soil but in forms that are not readily available to biota
- PO_4^{-3} is an available form of P
- P is important because ...

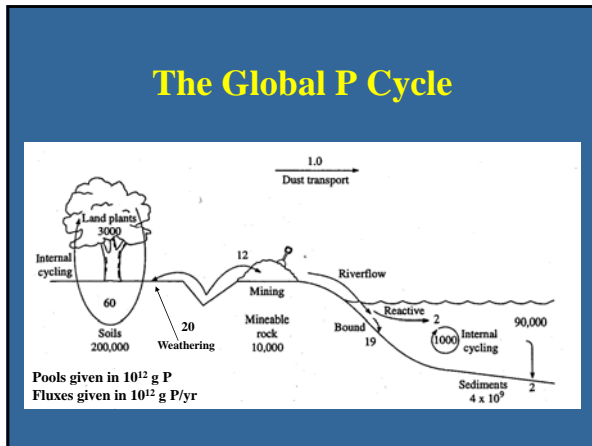
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- ### Summary of P Cycle
- Abundant but low availability.
 - Weathering of P-rich rock is original source.
 - Geologic processes are slow (millions of years) so biota rapidly recycle organic-P.
 - Residence time in biota is only a few days in the ocean.

- ### Summary of P Cycle
- Large loss to ocean relative to rate of return to land.
 - Losses in runoff are 93% particulate-P
 - Mycorrhizae ↑ absorption by plant roots
 - Mining P-rich rocks is a major source to land.
